

## A Level Chemistry Curriculum Map

**Teacher 1** -Module 2 atoms and bonding, module 4, module 6 and module 5 transition elements

**Teacher 2**- Module 2 amount of substance and acids and redox, module 3 and module 5 (excluding module 5 transition elements)

**Unit 1 Practical skills and techniques are taught alongside the theory covered in modules 2-6**

	Term 1	Term 2	Term 3	Term 4	Term 5	Term 6
Year 12	<ul style="list-style-type: none"> <li>Transition from GCSE to A level</li> </ul> <p>Module 1 PAG 1 – Moles determination</p> <p>Module 2           <ul style="list-style-type: none"> <li>Atoms, ions and compounds</li> <li>Amount of Substance</li> <li>Electrons and bonding</li> <li>Shapes of molecules and intermolecular forces</li> </ul> </p>	<p>Module 1 PAG 2 – Acid-base titration</p> <p>Module 2           <ul style="list-style-type: none"> <li>Acids and redox</li> </ul> </p> <p>Module 3           <ul style="list-style-type: none"> <li>Periodic table and energy</li> </ul> </p> <p>Module 4           <ul style="list-style-type: none"> <li>Basic concepts of organic chemistry</li> <li>Alkanes</li> <li>Alkenes</li> </ul> </p>	<p>Module 1 PAG 4 – Qualitative analysis of ions</p> <p>Module 3           <ul style="list-style-type: none"> <li>Reactivity trends</li> <li>Enthalpy changes</li> </ul> </p> <p>Module 4           <ul style="list-style-type: none"> <li>Alcohols</li> <li>Haloalkanes</li> </ul> </p>	<p>Module 3           <ul style="list-style-type: none"> <li>Enthalpy cycles</li> <li>Reaction rates and equilibria</li> </ul> </p> <p>Module 4           <ul style="list-style-type: none"> <li>Synthetic routes</li> <li>Spectroscopy</li> </ul> </p>	<p>Yr 12 Assessments</p> <p>Module 1 PAG 3 – determination of enthalpy changes</p> <p>PAG 5 – Synthesis of an organic liquid</p> <p>Module 3           <ul style="list-style-type: none"> <li>Measuring enthalpy change</li> </ul> </p> <p>Module 4           <ul style="list-style-type: none"> <li>Synthetic techniques (organic liquids)</li> </ul> </p>	<p>Module 1 PAG 6 – Synthesis of an organic solid</p> <p>PAG 9 – rates of reaction – continuous monitoring method</p> <p>PAG 10 – rates of reaction – initial rates method</p> <p>Module 5           <ul style="list-style-type: none"> <li>Rates of reaction and the rate equation</li> </ul> </p> <p>Module 6           <ul style="list-style-type: none"> <li>Aromatic Chemistry</li> <li>Synthetic techniques (organic solids)</li> </ul> </p>
Year 13	<p>Module 5           <ul style="list-style-type: none"> <li>Equilibrium constants</li> <li>Acids, bases and pH</li> </ul> </p> <p>Module 6           <ul style="list-style-type: none"> <li>Carbonyls and carboxylic acids</li> <li>Amines, amino acids and proteins</li> </ul> </p>	<p>Mock exams</p> <p>PAG 11 pH measurement</p> <p>Module 5           <ul style="list-style-type: none"> <li>Buffers and neutralisation</li> </ul> </p> <p>Module 6           <ul style="list-style-type: none"> <li>Organic synthetic pathways</li> </ul> </p>	<p>Module 1 PAG 7 – Qualitative analysis of organic functional groups</p> <p>PAG 12 – Research skills</p> <p>Module 5           <ul style="list-style-type: none"> <li>Enthalpy and entropy</li> <li>Redox</li> </ul> </p> <p>Module 6           <ul style="list-style-type: none"> <li>Chromatography and spectroscopy</li> </ul> </p>	<p>Mock exams</p> <p>Module 1 – PAG 8 - Electrochemical cells</p> <p>Module 5           <ul style="list-style-type: none"> <li>Electrode potentials</li> <li>Transition elements</li> </ul> </p>	<p>Revision, consolidation and exam preparation</p>	<p>External A Level Exams</p>